Cobalt Chloride Thermometer - Worksheet

Name...... Class.....

1. The structures of the pink and blue compounds are shown below. What is the geometry of the molecule around the metal centre?

$$H_2O/I_{I_{I_1}}$$
 OH_2
 OH_2
 OH_2
 OH_2
 OH_2
 OH_2
 OH_2
 OH_2
 OH_3
 OH_4
 OH_4
 OH_5
 OH_6
 OH_6
 OH_7
 OH_8
 OH

Geometry = Geometry =

- 2. By changing the temperature of the solution, you are changing the position of equilibrium of the reaction and therefore the colour of the solution. Mark with an arrow on the diagram below the position of equilibrium when the solution is....
 - a. Hot?
 - b. Cold?
 - c. Ambient (room temperature)

3. Give an example of an everyday use of colour thermometers.....